

Microwave and Millimeterwave Spectrum of Monofluoracetonitrile- ^{15}N ($\text{CH}_2\text{FC}^{15}\text{N}$). A Contribution to Molecular Structure

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The microwave and millimeterwave spectrum of ^{15}N -fluoracetonitrile ($\text{CH}_2\text{FC}^{15}\text{N}$) in the ground vibrational state are investigated in the region between 8 and 150 GHz. The measured transitions are fitted to a Hamiltonian with three rotational constants, five quartic and seven sextic centrifugal distortion constants in the symmetric top limit of van Eijck-Typke and in the s-reduction of Watson. The rotational constants are used to improve the calculated r_0 -structure of the fluoracetonitrile molecule.

Introduction

The microwave spectra of the fluoracetonitrile isotopomers CH_2FCN , CHDFCN and CD_2FCN have been investigated by Kasten, Dreizler, Job and J. Sheridan [1]. These authors considered low J lines in the frequency range up to 30 GHz only. A further investigation in the millimeter range has been performed in this laboratory [2] to get detailed informations about quartic and sextic centrifugal distortion terms. In the present work, following similar criteria as in [2], the millimeterwave and microwave spectrum of $\text{CH}_2\text{FC}^{15}\text{N}$ has been investigated and 123 lines have been measured and assigned. These lines have been fitted using both the Hamiltonian of van Eijck-Typke and of Watson's symmetric top reduction. Rotational constants as well as quartic and sextic centrifugal distortion constants have been determined. In reference [1] a first approach to a r_0 -structure had been made taking in account all rotational constants available. Additional data from the present investigation have been used with the purpose of improving the structure parameters and to give better initial constants for the calculation of the spectra of two other isotopomers with ^{13}C which are being investigated.

Experimental

The ^{15}N -fluoracetonitrile was prepared by dehydration of ^{15}N -fluoracetamide with phosphorpent-

oxide [3]. The ^{15}N -fluoracetamide is easily prepared from ^{15}N -ammonia and ethylfluoracetate [4]. Most of the lines were measured using source modulation. A 16.7 kHz sine wave was used to modulate the standard frequency which stabilizes the klystron. After detection the modulated signal was amplified in a narrow band amplifier at 33.4 kHz and phase detected. The frequency sweep was provided by an externally controlled ramp voltage which drives the frequency decade Schomandl ND 800. The frequencies were stabilized by a Schomandl FDS 30 synchronizer. For these measurements the absorption cell was a glass cell with an inner diameter of 10 cm and a length of 150 cm.

The transitions were assigned on the basis of an 'a priori' calculation and lines pattern comparison with the normal isotopomer. In some special cases the assignment has been made using Stark modulation or microwave-millimeterwave double resonance technique (MW-MMW-DR) [5]. For the measurements with Stark modulation and MW-MMW-DR technique a conventional Stark waveguide cell was used. The microwave and millimeterwave frequencies were produced by different OKAYA klystrons either directly or by harmonic multiplication of the fundamental frequency.

All frequency measurements have been carried out using as standard frequency a 5 MHz signal derived from the 77.5 kHz signal of the broadcasting station DCF 77 (Mainflingen, Germany) with a relative accuracy of $5 \cdot 10^{-13}$.

The measurements of the lines are believed to be better than $1 \cdot 10^{-7}$.

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Table 4. Watson's determinable parameters obtained from both sets of centrifugal distortion constants. Errors, given in parenthesis in units of the last digit, are derived from standard errors of the least squares fitting procedure as in Table 2.

	Watson's s-reduction	van Eijck-Typke	
<i>A</i>	36417.7434(1040)	36417.7434(1040)	MHz
<i>B</i>	4626.49037(438)	4626.49037(438)	MHz
<i>C</i>	4209.68743(405)	4209.68743(407)	MHz
τ_{AAAA}	21.390(438)	21.390(436)	MHz
τ_{BBBB}	−14.311(360)	−14.308(355)	kHz
τ_{CCCC}	−6.058(238)	−6.063(239)	kHz
τ_1	237.59(84)	237.59(83)	kHz
$\tau_2/(A+B+C)$	15.259(634)	15.258(633)	Hz
Φ_{AAA}	−19.63(2168)	−19.61(2170)	Hz
Φ_{BBB}	−0.299(236)	−0.137(234)	Hz
Φ_{CCC}	−0.159(156)	−0.320(176)	Hz
Φ_1	−11.47(306)	−11.47(387)	Hz
$\Phi_2 + \Phi_3/(A+B+C)$	−0.215(633)	−0.212(544)	Hz
$\Phi_2 - \Phi_3/(A+B+C)$	0.766(224)	0.652(544)	Hz
$\Phi_4/(A+B+C)$	−1.252(52)	−1.252(51)	kHz

Results

Table 1 gives only a small part of the lines measured. A complete list of lines can be obtained from the author or the Central Library of the University of Kiel under the number TNA 10. The frequencies have been fitted to the symmetric top limit Hamiltonian of van Eijck [6] and Typke [7]

$$\begin{aligned}
 H' = & B'_x P_x^2 + B'_y P_y^2 + B'_z P_z^2 - D'_J (P^2)^2 - D'_{JK} P^2 P_z^2 \\
 & - D'_K P_z^4 - \delta_J [P^2 (P_x^2 - P_y^2) + (P_x^2 - P_y^2) P^2] \\
 & - 2R'_6 [3(P_x^2 P_y^2 + P_y^2 P_x^2) - P_x^4 - P_y^4] + H'_J (P^2)^3 \\
 & + H'_{JK} (P^2)^2 P_z^2 + H'_{KJ} P^2 P_z^4 + H'_K P_z^6 \\
 & + H'_5 (P^2)^2 (P_x^2 - P_y^2) \\
 & + \frac{1}{2} H'_6 [3(P_x^2 P_y^2 + P_y^2 P_x^2) - P_x^4 - P_y^4] \\
 & + H'_{10} (P_x^2 - P_y^2)
 \end{aligned}$$

and to the Watson's s-reduced Hamiltonian [8]

$$\begin{aligned}
 H^{(s)} = & \frac{1}{2} (B_x^{(s)} - B_y^{(s)}) P^2 + [B_z^{(s)} - (B_x^{(s)} + B_y^{(s)})/2] P_x^2 \\
 & + \frac{1}{4} (B_x^{(s)} - B_y^{(s)}) (P_+^2 + P_-^2) - D'_J (P^2)^2 \\
 & - D'_{JK} P^2 P_z^2 - D'_K P_z^4 - d_1 P^2 (P_+^2 + P_-^2) \\
 & + d_2 (P_+^4 + P_-^4) + H'_J (P^2)^3 + H'_{JK} (P^2)^2 P_z^2 \\
 & + H'_{KJ} P^2 P_z^4 + H'_K P_z^6 + h_1 (P^2)^2 (P_+^2 + P_-^2) \\
 & + h_2 P^2 (P_+^2 + P_-^2) + h_3 (P_+^6 + P_-^6);
 \end{aligned}$$

$$(P_{\pm} = P_x \pm i P_y).$$

To fit the rotational and centrifugal distortion constants the computer program ZFAP6 (author: Typke) has been used. The resulting parameters are listed in

Table 2. The corresponding correlation matrices are listed in Table 3a and 3b. For comparison, Watson's determinable parameters were calculated from both sets of constants [9] (see Table 4).

Fitting of the Structure

A preliminary r_0 structure of CH₂FCN has already been given in [1], where several parameters were fixed to literature values. With the three rotational constants of CH₂FC¹⁵N we should be able to give a r_0 -structure using a smaller number of fixed structural parameters. For the fit the rotational constants of CH₂FC¹⁴N, CHDFCN, CD₂FCN and CH₂FC¹⁵N have been used. The rotational constants of the double hydrogen substituted isotopomer, CD₂FCN, really give no further informations on the coordinates of the hydrogen atom, nevertheless they have been used as additional secondary information. It was not possible to fit more than five structure parameters with these twelve constants, so that $r(\text{C-N})$ [10–13] and $r(\text{C-C})$ [10, 12–14] or $r(\text{C-F})$ [15–17] were assumed

Table 5. Partial r_0 -structure of CH₂FCN calculated from twelve rotational constants. Assumptions in brackets. a) $r(\text{C-C})$ and $r(\text{C-N})$ have been held fixed as in [1]. b) $r(\text{C-C})$ and $r(\text{C-F})$ have been held fixed, taking in account that substitution of H and N gives informations on the coordinates of these nuclei. c) $r(\text{C-N})$ and $r(\text{C-F})$ have been held fixed. The last three angles are derived from the fitted values. Errors given in units of the last digit are obtained from the least squares fitting procedure or are deviations from the mean value of the parameters assumed from related compounds.

	a)	b)	c)
$r(\text{C-N})$	(1.158(1)A)*	1.15(9)A	(1.158(1)A)*
$r(\text{C-C})$	(1.460(1)A)**	(1.460(1)A)**	1.45(8)A
$r(\text{C-H})$	1.09(25)A	1.08(23)A	1.09(26)A
$r(\text{C-F})$	1.38(6)A	(1.381(7)A)***	(1.381(7)A)***
$\angle \text{HCC}$	110.8(6)°	109.2(7)°	110.6(6)°
$\angle \text{FCC}$	111.2(0)°	111.3(0)°	111.3(0)°
$\angle \text{CCN}$	(180°)	(180°)	(180°)
$\angle \text{CCF/CCH}$	121.3(2)°	121.6(2)°	121.3(2)°
$\angle \text{CCH/CCH}$	117.3°	116.7°	117.3°
$\angle \text{HCH}$	106.0°	107.1°	106.2°
$\angle \text{CN, a}$	19.6°	19.8°	19.8°

* Average value from CH₃CN [10], (CH₃)₃CCN [11], CCl₃CN [12] and NC-CH₂-CN [13].

** Average value from CH₃CN [10], CCl₃CN [12], NC-CH₂-CN [13], and CF₃CN [14].

*** Average value from CH₃F [15], CH₂BrF [16], CH₂CIF [17].

from known geometries of related molecular structures. The CCN group was supposed to be linear. For comparison a fit with six or more free parameters has been carried out. The correlation values between the constants were then extremely high and the resulting atomic distances and angles were very different from the values of comparable molecules. Therefore only five structure parameters have been fitted and the result is listed in Table 5. The spectrum of two ¹³C

isotopomers is now under investigation in order to give a better *r*₀-structure.

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